

# Investigation of ligand binding mode at 5-HT<sub>6</sub>R with the use of bioisosterism

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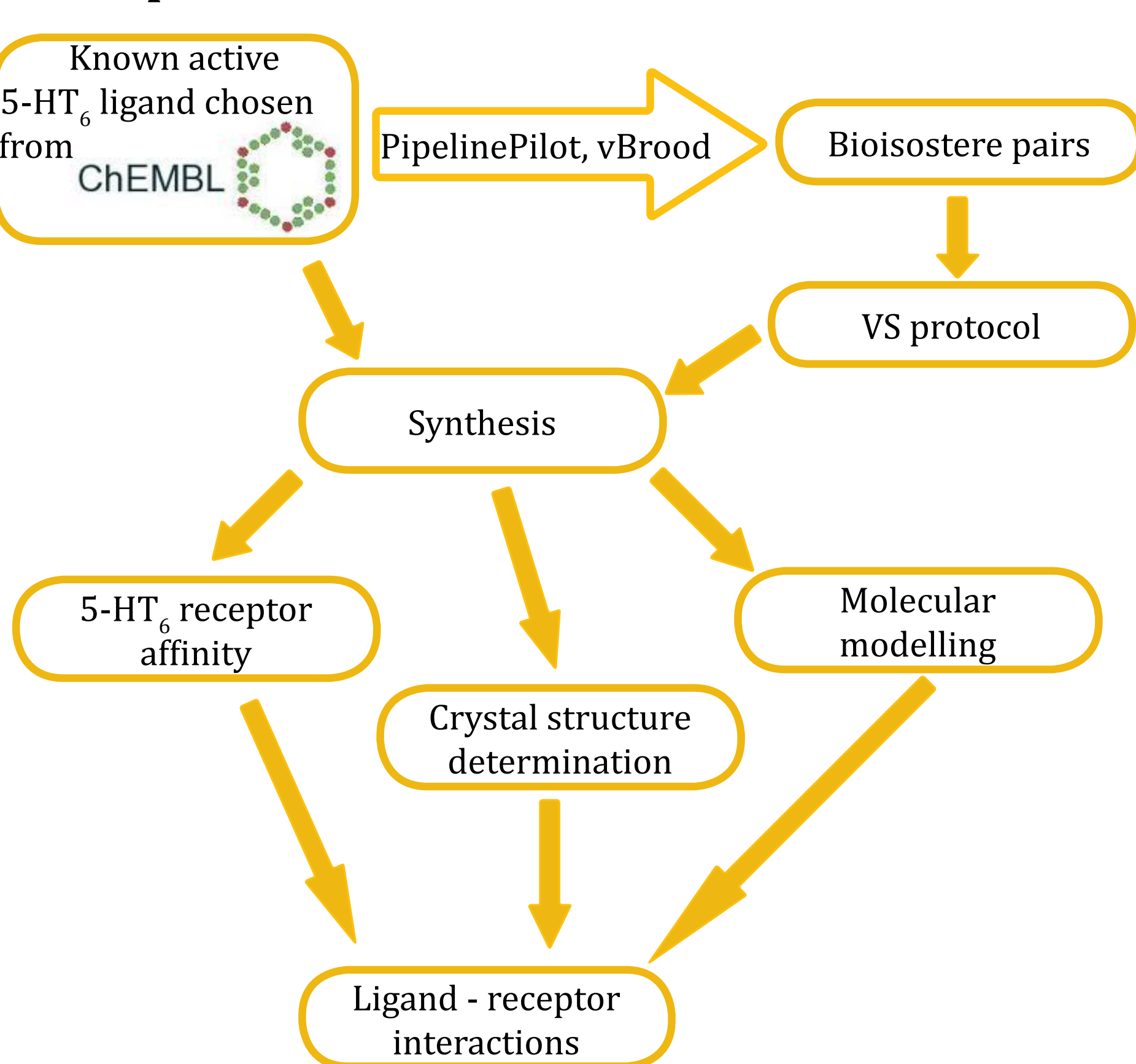
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## Introduction

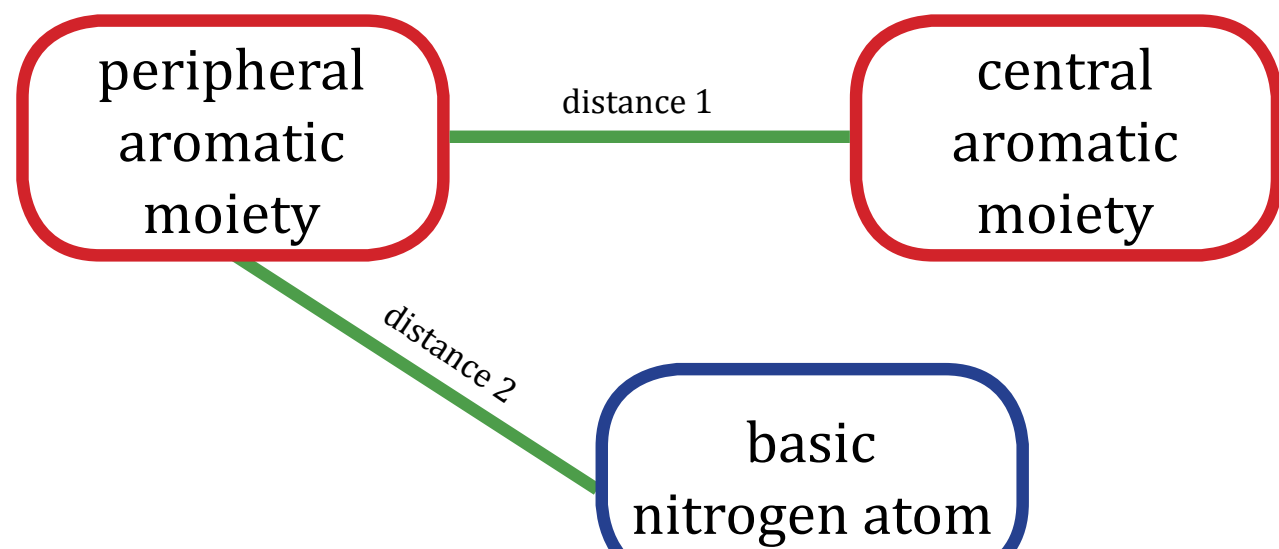
One of the most recently identified serotonin receptor subtypes – the 5-HT<sub>6</sub> receptor, localized practically only in the brain, is a very promising target for different new psychotropic drugs.[1,2,3] These receptors are supposed to be responsible mainly for motor control, memory and learning and its ligands can be used to treat cognitive impairments and also as antiobesity drugs.[4-8] So far, several thousand of ligands have been synthesized and their structural diversity makes consensus binding mode very difficult to be defined. Isosterism is the most common technique used by medicinal chemists to design and synthesize new series of compounds. An isosteric replacement can change compound activity, bioavailability, pharmacokinetics and metabolism. If isosteric replacement doesn't substantially change biological properties of a substance, it is called bioisosteric replacement. Besides altering compound properties, bioisosterism can be used to get insight into interactions of ligand with the receptor. By carefully planning isosteric replacements it is possible to probe certain regions of receptor binding pocket.

## Concept



## Crystallographic studies

Up to date crystal structures of several analysed compounds were obtained. Two distances in crystal structures were measured and compared: between central and peripheral aromatic moiety and between peripheral aromatic moiety and basic nitrogen atom.



Compound	distance 1 [Å]	distance 2 [Å]
1	5.332	9.138
2	6.452	9.785
6	5.448	8.146
7	6.540	7.344
8	6.01	7.829
10	5.059	6.771
11	6.209	10.004
12	6.427	9.956
13	5.524	7.335
21	6.633	8.527
average		
active (K <sub>i</sub> < 100 nM)	5.792 ± 0.488	7.870 ± 0.751
inactive (K <sub>i</sub> > 100 nM)	6.363 ± 0.109	9.915 ± 0.094

## Docking studies

In order to measure the position of a ligand in a binding pocket of 5-HT<sub>6</sub>R, distances between ligand and different amino acid residues were calculated for 100 best scored complexes.

Anchor points for measurement		Average distance [Å] for <b>active</b> ligands ( $K_i < 100$ nM)	Average distance [Å] for <b>inactive</b> ligands ( $K_i > 100$ nM)
Amino acid residue	Ligand		
W6.48	Nearest ligand atom	6.220 ± 1.185	6.572 ± 1.325
Nearest aa. in a loop between 2 <sup>nd</sup> and 3 <sup>rd</sup> helix	Nearest ligand atom	2.021 ± 0.182	2.016 ± 0.204
	Ionised basic nitrogen atom	6.832 ± 3.874	6.320 ± 3.437
F6.51	Nearest aromatic moiety	5.859 ± 0.816	6.735 ± 1.056
F6.52	Nearest aromatic moiety	6.623 ± 1.079	7.202 ± 1.807

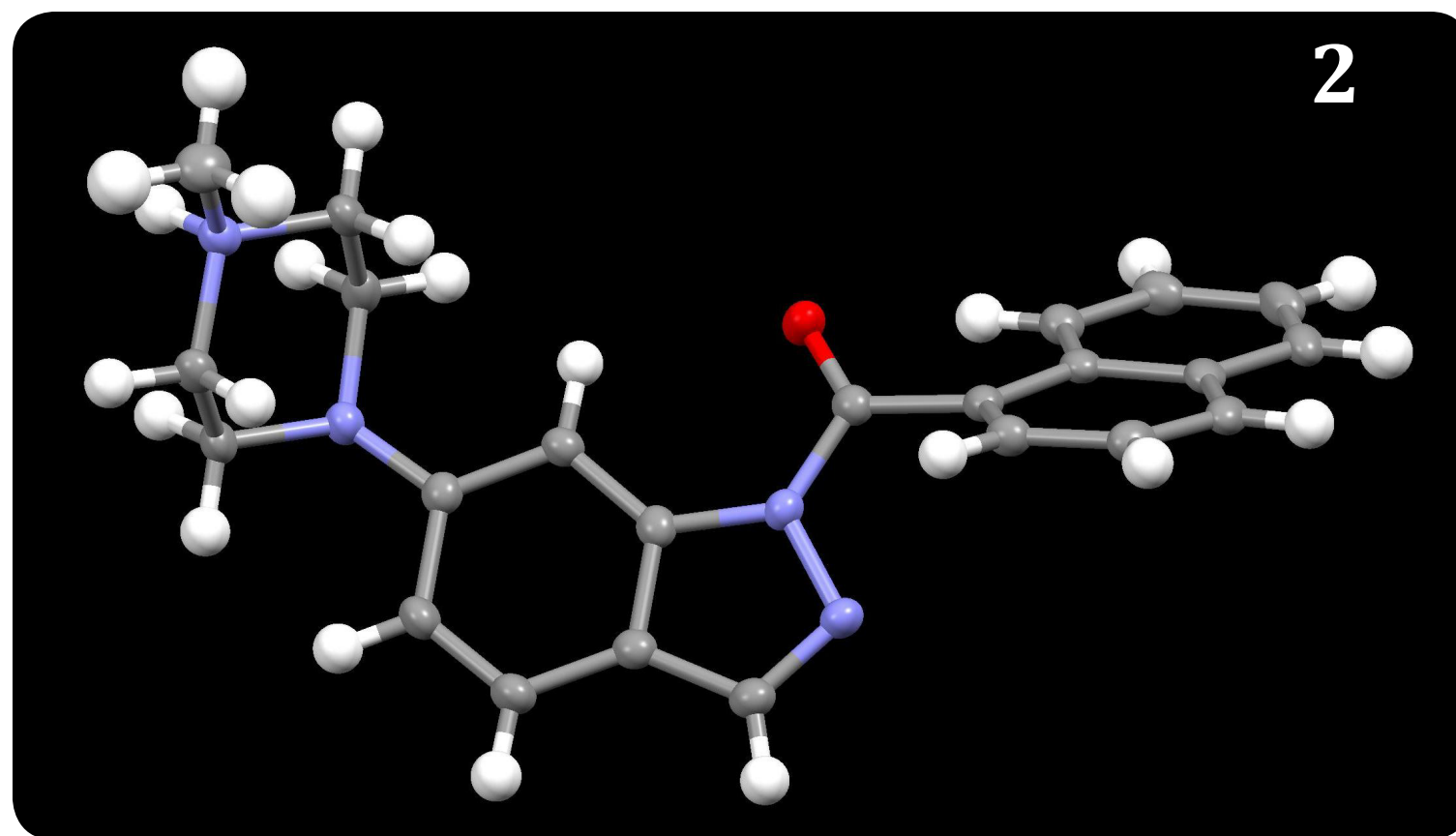
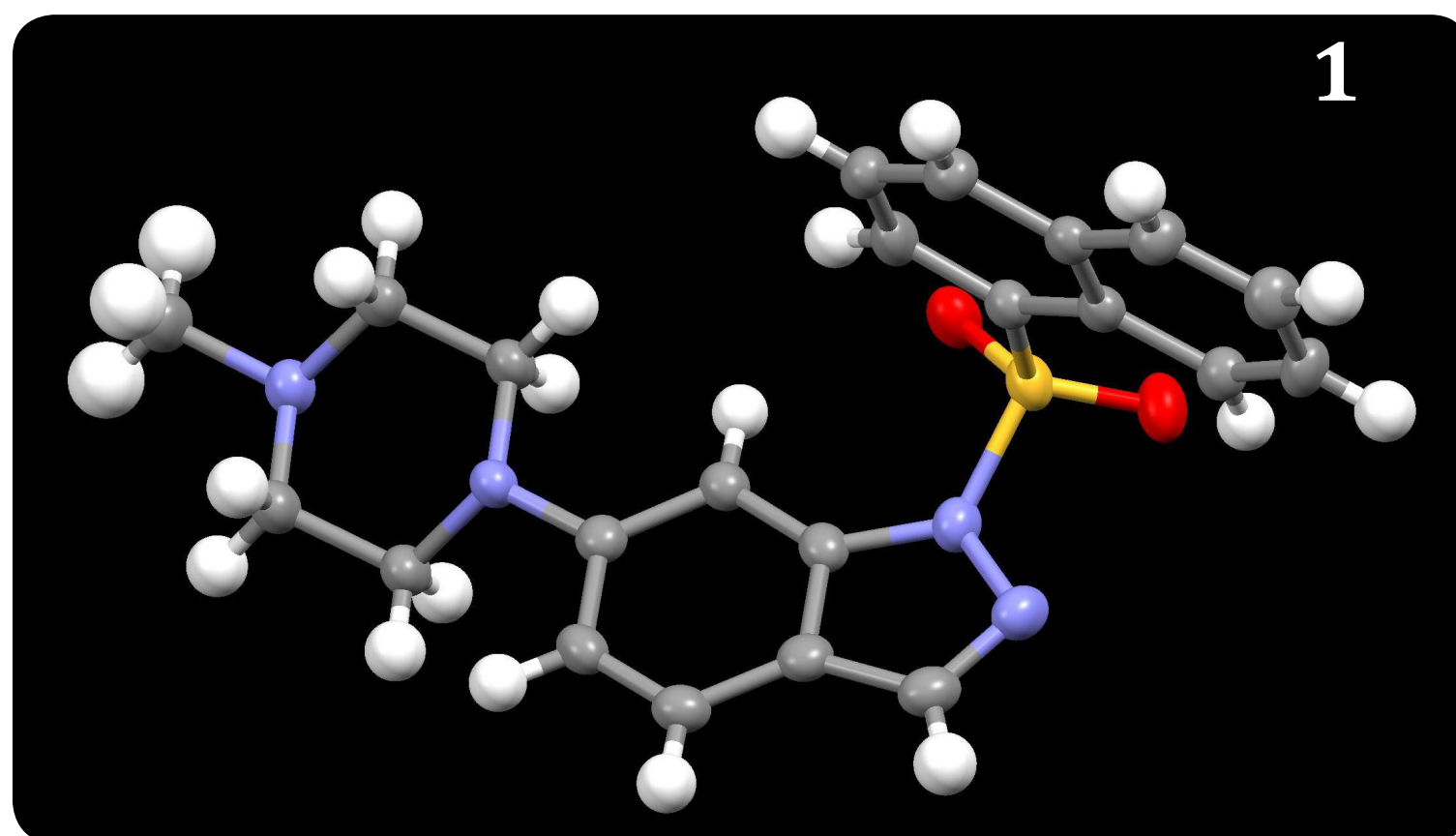
## SIFt representation

For each compound, only the best docking pose per receptor model was considered and 100 the best scored complexes were transformed into bitstring applying SIFt formalism statistically describing interactions between ligand and receptor.[9,10]

Residue	[%] compounds interacting with residue		average SIFt	
	active K <sub>i</sub> < 100 nM	inactive K <sub>i</sub> > 100 nM	active K <sub>i</sub> < 100 nM	inactive K <sub>i</sub> > 100 nM
W3.28	42	50	0.69	0.72
T3.29	83	75	0.68	0.62
D3.32	100	83	0.88	0.78
V3.33	100	100	0.90	0.83
C3.36	100	83	0.81	0.76
L4.61	50	50	0.62	0.61
G146	0	25	-	0.60
R162	0	17	-	0.57
L163	100	92	0.98	0.92
L164	67	67	0.71	0.76
A165	100	92	0.82	0.74
F5.38	100	100	0.81	0.76
V5.39	100	100	0.85	0.83
A5.42	92	83	0.72	0.70
S5.43	33	50	0.82	0.64
T5.46	75	42	0.61	0.65
W6.48	75	50	0.64	0.57
F6.51	100	100	0.99	0.95
F6.52	100	83	0.79	0.73
N6.55	100	92	0.85	0.82
V6.58	58	17	0.66	0.71
F7.35	100	100	0.99	0.99
D7.36	17	50	0.62	0.72
T7.39	100	100	0.92	0.84
Y7.43	100	83	0.73	0.70

## Crystal structures

Crystal structures of **1** and **2**. Noticable is different mutual orientation of both aromatic moieties.



## Conclusions

The goal of research was to investigate ligand-receptor interactions using designed and synthesized bioisosteric pairs. As a result, several amino acids from receptor binding pocket were highlighted as significant for binding ligands with high affinity for 5-HT<sub>6</sub> receptor. These amino acids are: D3.32, V3.33, C3.36, S5.43, T5.46, W6.48, F6.52 and V6.58. Especially, the phenylalanine cluster of F6.51 and F6.52 was selected as most important due to statistically closer position of potent ligands to these amino acids.

Additionally, crystal structures revealed significant differences in intermolecular distances between active and inactive compounds, being much shorter for ligands with high affinity.

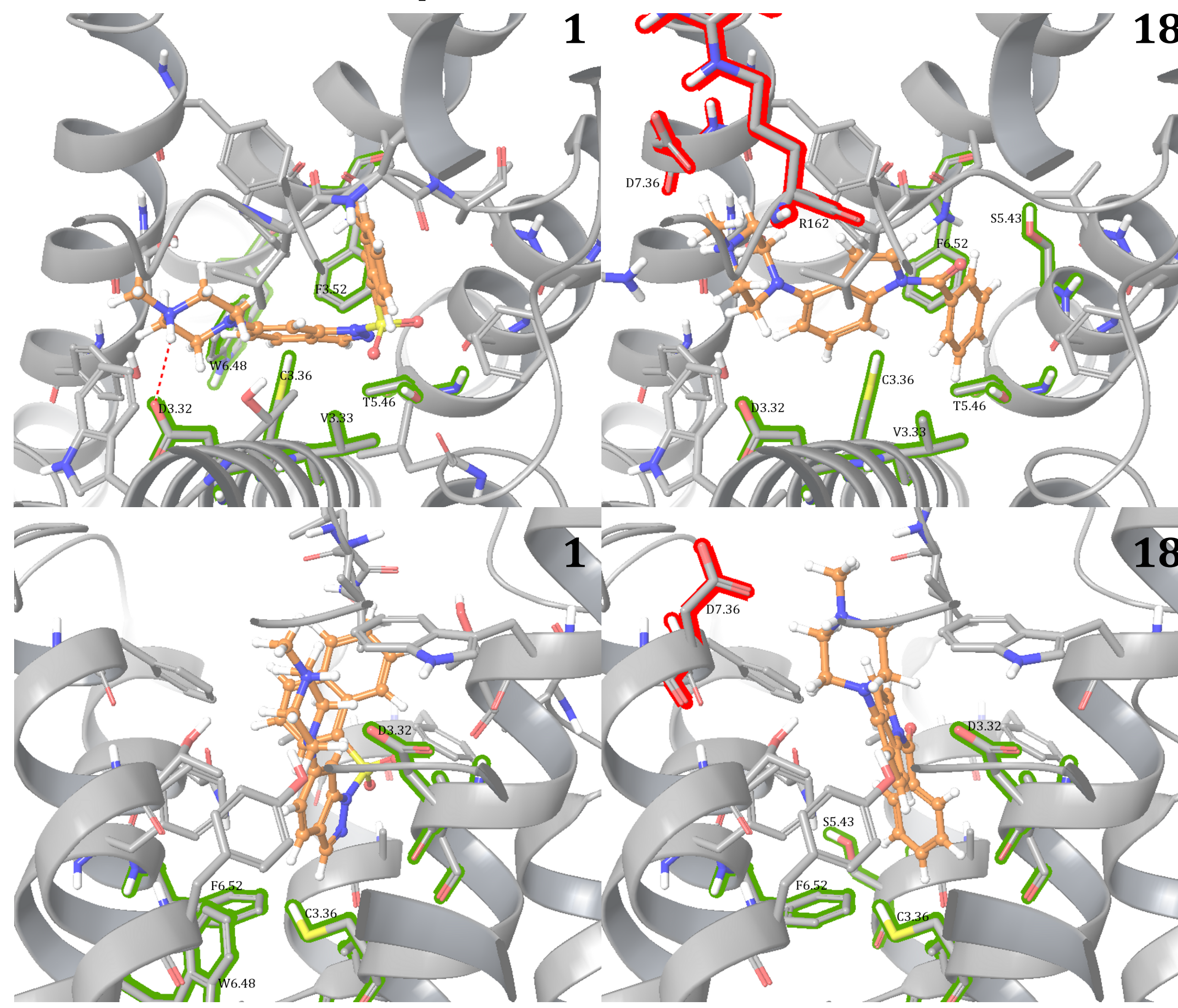
Conducted research provided useful introduction to additional studies of the ligand-receptor interactions based on wider group of compounds. What more, developed methodology can be implemented to examine interactions of ligands with another receptors.

## Bioisosteric pairs Affinity values (K<sub>i</sub>) for 5-HT<sub>6</sub>R in [nM] measured in our laboratory.

Ligand	Bioisostere	Type of substitution
1 (K <sub>i</sub> = 1)	2 (K <sub>i</sub> = 1280)	sulphonyl - carbonyl
3 (K <sub>i</sub> = 1)	4 (K <sub>i</sub> = 2067)	sulphonyl - carbonyl, methanediyl
5 (K <sub>i</sub> = 116)		
6 (K <sub>i</sub> = 11)	7 (K <sub>i</sub> = 44)	sulphonyl - carbonyl, methanediyl
8 (K <sub>i</sub> = 23)	9 (K <sub>i</sub> = 202)	
10 (K <sub>i</sub> = 22)	11 (K <sub>i</sub> = 245)	noncyclic - cyclic ring expanding
12 (K <sub>i</sub> = 62)	13 (K <sub>i</sub> = 6)	
14 (K <sub>i</sub> = 4)	15 (K <sub>i</sub> = 187)	sulphonyl - carbonyl, methanediyl
16 (K <sub>i</sub> = 18)		
17 (K <sub>i</sub> = 1)	18 (K <sub>i</sub> = 2204)	sulphonyl - carbonyl, methanediyl
19 (K <sub>i</sub> = 24)		
20 (K <sub>i</sub> = 22)	21 (K <sub>i</sub> = 63)	cyclic - noncyclic
22 (K <sub>i</sub> = 2280)	23 (K <sub>i</sub> = 2675)	phenyl - cyclohexane
	24 (K <sub>i</sub> = 3760)	piperidine - phenyl

## Receptor-Ligand Complexes

Representative complexes with one of receptor conformation for active (**1**) and inactive (**18**) compounds. Aminoacid residues selected as important for active ligands interaction with receptor are marked in green. Aminoacid residues responsible for interaction with inactive compounds are marked in red.



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